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UNIT 5 SPECIMEN
MARK

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SCHEME/VERSION 1.1

4 (c) (i) For a reaction to occur $\Delta G < 0$ ΔS is

positive and large as a gas is evolved $T \Delta S$ is larger

than ΔH and ΔG is negative (1) (1) (1) (ii)

ΔS is negative Four moles gaseous reactant forming or more moles of gaseous product

GCE Chemistry Unit 5
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GCE Chemistry Mark
Scheme Unit 05 -
Energetics, Redox and
Inorganic Chemistry
June 2012. Version 1.1.
General Certificate of
Education (A-level) June
2012. Chemistry
(Specification 2420)
CHEM5 Unit 5:
Energetics, Redox and
Inorganic Chemistry.
Final. Mark Scheme.
Mark schemes are

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5 Mark Scheme
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prepared by the Principal
Examiner and
considered, together with
the relevant questions, by
a panel of subject
teachers.

GCE Chemistry Mark
Scheme Unit 05 -
Energetics, Redox and ...
Chemistry (CHM5) -
AQA GCE Mark
Scheme 2007 June series
5 (c) $K_a = [H$

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$$+][A^-]/[HA] \text{ or } =$$
$$[H^+]^2/[HA] \quad (1) \quad [H^+] =$$
$$K_a[HA] =$$

$$(1.15 \times 10^{-4} \times 0.5)$$

(mark is for expression
or numbers) (1)

$$= 7.58 \times 10^{-3} \text{ mol dm}^{-3}$$

(1) $\text{pH} = -\log_{10}[H^+]$
(or log or lg) (allow last
two marks consequential
on wrong $[H^+]$) (1)

GCE Chemistry Unit 5
Mark Scheme June 2007

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(or = $4.2(5) \times 10^3$ if
candidate uses given
moles of hydrogen) 1

Mass of iron = $4.53 \times$
 $10^3 \times 55.8 = 0.253$ g
(mark is for method mass
= moles \times A r) (Mass of
iron can be 56) 1 (iv)
 $0.253 \times 100/0.263 =$
96.1 % (mark is for
answer to 2 sig. figs.)

(allow conseq on mass of
iron. E.g. = 90% from
 $4.2(5) \times 10^3$ moles of

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5 (H2 and Fe) Scheme
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GCE Chemistry Unit 5
Mark Scheme June 2006
International GCE
Chemistry (6CH05/01R)
Edexcel and BTEC
Qualifications ... mark
the first candidate in
exactly the same way as
they mark the last. Mark
schemes should be
applied positively.
Candidates must be ... =

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$$5.516 \times 10^{-4} \text{ (mol) (1)}$$
$$\% \text{ Oxidized} = 5.516 \times 10^{-4} \times 100 / 2 \dots$$

Mark Scheme (Results)
June 2014 - Edexcel
Cleared/Standard MS:
AS 2, Further Physical
and Inorganic Chemistry
and an Introduction to
Organic Chemistry -
[467] GCE Chemistry
(2016), Summer 2019
Archived Past Papers &

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5 Mark Scheme
Cleared/Standard MS:
A2 3, Further Practical
Chemistry (Booklet B)
(Theory) - [467] GCE
Chemistry (2016),
Summer 2019

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Summer 2013 GCE
Chemistry 6CH04/01
General Principles of

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Chemistry I. Scheme

6CH04_01_1306 Edexcel
and BTEC Qualifications

... Correct Answer Reject

Mark 5(a) B 1 (b) C 1 (c)

B 1 Question Number

Correct Answer Reject

Mark 6 A 1 Question

Mark Scheme (Results)

Summer 2013 - Edexcel

GCE Chemistry

(6CH04/01R) Edexcel

and BTEC Qualifications

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5... Reject Mark 5 C 1

Question Number

Correct Answer Reject

Mark 6(a) B 1 Question

number ... Value, sign

and unit (1) Ignore SF

except one Internal TE

allowed for recognisable

numbers, for example:

Ho at calcium ...

Mark Scheme (Results)

June 2014 - Edexcel

Unit-1 | Unit-2. Unit-1 |

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Unit-2A. Unit-1 |
Unit-2. Unit-1 | Unit-2B.
Unit-4A | Unit-4B.
Unit-4C | Unit-4D.
Unit-1 | Unit-2. Unit-1 |
Unit-2. Unit-4 | Unit-5.

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2008) May/June 2017
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Chemistry specification
provides a broad,
coherent, satisfying and

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AS/A Level Chemistry
Title: GCE Chemistry
June 2003 Mark Scheme
Author: AQA Created

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Date: 11/19/2003 1:53:00
PM

June 2006

GCE Chemistry June
2003 Mark Scheme -
TomRed's Stuff

GCE CHEMISTRY
UNIT 4 SPECIMEN
MARK SCHEME /

VERSION 0.2 7 = 264 5

135. $\times 100 = 51.0 \% (1)$

(ii) expected yield = 93

10 $\times 0.5 \times 135 = 7.26$

kg % yield = 7 26 5 38..

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$$\times 100 = 74.1 \% (1) (1)$$

(iii) Although yield appears satisfactory (74%) % atom economy is only 51% QWC nearly half of the material produced is waste and must be disposed of QWC ...

GCE Chemistry Unit 4
Specimen Mark Scheme
GCE CHEMISTRY
UNIT 2 SPECIMEN

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MARK SCHEME/
VERSION 1.1 7

Question 7 (a) (i) Can
still score $\frac{3}{4}$ i.e. penalise
M3 3 (4) If wrong
carbocation, lose
structure mark If wrong
alkene, lose structure
mark Penalise M2 if
polarity included
incorrectly no bond
between H and Br bond
is shown as or (ii) CH
CH CH₃ 22(1) +

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GCE Chemistry Unit 2
Specimen Mark Scheme

1st mark: • top arrow must start from the double bond / close to the double bond and not from either of the C atoms of the $C=C$ bond

- top arrow can end on, or close to, the H in HBr
- lower arrow must start from the bond and not the H atom in HBr.

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3 Mark Scheme
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REJECT full charges on
the HBr. 2nd mark:

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June 2010 - Edexcel
Chemistry Version 1.0
AS exams 2009 onwards
A2 exams 2010 onwards
GCE AS and A Level
Unit 3X: EMPA
Specimen mark scheme

GCE Chemistry Unit 3X
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5 Mark Scheme (Results)

Summer 2013 GCE

Chemistry 6CH02/01R

... Chemistry . Edexcel
and BTEC Qualifications

Edexcel and BTEC

qualifications come from

Pearson, the world ' s

leading learning

company. ... Reject Mark

5 D 1 Question Number

Correct Answer Reject

Mark 6 A 1 Question

Number

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Mark Scheme (Results)
Summer 2013 - Edexcel

GCE Chemistry A Unit
H032/01: Breadth in
chemistry Advanced
Subsidiary GCE Mark
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Cambridge and RSA) is a
leading UK awarding
body, providing a wide
range of qualifications to
meet the needs of

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5 Mark Scheme
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candidates of all ages and
abilities. OCR
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GCE Chemistry A - Past
Papers

Title: GCE Chemistry
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