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Examples, Ch. 21: 11-14, 17-19, 21, 24, 27-30, 59-60, 73 II. Rates of Decay (21.4): Know the half-life equation for radioactive decay (first order reaction) and be able to use it to convert between amount and time, or vice versa. Examples, Ch. 21: 34-42, 80 III. Energy Changes (21.6): Know how to calculate the energy change for a nuclear reaction

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### Chapter 15 An Introduction to Organic Chemistry ...

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Chemistry Honors Ch ( 1-7.1 )-Midterm study Guide Choose the best answer from the options that follow each question.   A 1. A scientist is working in the lab, with compounds containing carbon. She is most likely working in the field of The answer is A because organic chemistry is the study of most carbon-containing compounds.   A    2.

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